(12) UK Patent Application (19) GB (11) 2 003 477 A

- (21) Application No 7835369
- (22) Date of filing 1 Sep 1978
- (23) Claims filed 1 Sept 1978
- (30) Priority data
- (31) 10735/77 737/78
- (32) 2 Sep 1977 24 Jan 1978
- (33) Switzerland (CH)
- (43) Application published
- 14 Mar 1979 (51) INT CL² CO7D 231/64
- (52) Domestic classification C2C 1405 213 247 250 252 25Y 305 30Y 351 352 366 367 368 387 625 628 678 760 TY
- (56) Documents cited None
- (58) Field of search C2C
- (71) Applicant
 Ciba-Geigy AG,
 4002 Basie,
 Switzerland,
- (72) Inventor
 Anthanassios Tzikas
- (74) Agent Messrs. J.A. Kemp and Co.

(54) Process for the production of 3-substituted pyrazolanthrones

(57) A process for the production of 3-substituted pyrazolanthrones of the formula

wherein X is c_1 - C_6 alkyl, C_1 - C_6 alkoxy or a group -COR, in which R is hydrogen, C_1 - C_4 alkyl, hydroxyl, C_1 - C_6 alkoxy, -NH₂, -NH- C_1 - C_6 alkyl or -NR₁R₂, in which R₁ and R₂, each independently of the other, are C_1 - C_6 alkyl, which comprises reacting 1-nitroanthraquinones of the formula

wherein X is as defined in formula (1), in aprotic dipolar solvents, with hydrazine or hydrazine hydride, at normal or slightly elevated temperature, to give 3-substituted pyrazolanthrones of the formula (1), which we dyestuff intermediates

Process for the production of 3-substituted pyrazolanthrones

The present invention relates to a process for the production of 3-substituted pyrazolanthrones of the formula

wherein X is C₁-C₅alkyl, C₁-C₅alkoxy or a group -COR,
15 in which R is hydrogen, C₁-C₅alkyl, hydroxyl, C₁C₅alkoxy, -NH₂, -NH-C₁-C₅alkyl or -NR₁R₂, in which R₁
and R₂, each independently of the other, are C₁C₅alkyl, which comprises reacting
1-nitroanthraquinones of the formula

20

- 25 wherein X is as defined in formula (1), in aprotic dipolar solvents, with hydrazine or hydrazine hydride, at normal or slightly elevated temperature, to give 3-substituted pyrazolanthrones of the formula (1).
- 30 The substituent X is formulae (1) and (2) can be for example: methyl, ethyl, propyl, isopropyl, butyl, hexyl, methoxy, ethoxy, propoxy, isopropoxy, butoxy, formyl, acetyl, propionyl, carboxyl, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, carbamoyl, N-methylcarbamoyl, N-ethylcarbamol, N,N-dimethylcarbamoyl and N,N-diethylcarbamoyl.

As starting compounds of the formula (2), there may be mentioned in particular: 1-nitro-2-methyllanthraquinone, 1-nitro-2-ethylanthraquinone and 1-nitro-2-carboxyanthraquinone.

Examples of suitable aprotic dipolar solvents which may be used in the process of the invention are: N-methyl-2-pyrrolidone, tetramethylurea, sulfolane, hexamethylphosphoric triamide, dimethyl sulfoxide, dimethyl acetamide, diethyl acetamide, acetonitrile, dimethyl formamide, 3,3'-thiodipropionitrile and also pyridine.

The most advantageous reaction temperature depends on the starting compound and the solvent employed. Thus the reaction can often be carried out at room temperature. During the reaction, the temperature of the reaction mixture generally rises to 30° to 40°C. The reaction can also be initiated above room temperature (e.g. at 30°C) or below it (e.g. at 5° to 10°C). The advantageous temperature range for the process of the invention is that between about 0° and 100°C.

A preferred embodiment of the process of the invention consists in reacting 1-nitro-2-methyl-60 anthraquinone, 1-nitro-2-ethylanthraquinone or 1-nitro-2-carboxyanthraquinone in N-methyl-2-pyrrolidone, sulfonlane or dimethyl sulfoxide, with hydrazine hydrate, to give 3-methylpyrazolanthrone. 3-ethylpyrazolanthrone or 3-carboxy-

In the course of the process of the present invention, there is probably formed as intermediate the corresponding 2-substituted

1-hydrazinoanthraquinone, which is immediately 70 cyclised to the pyrazolanthrone.

The 3-substituted pyrazolanthrones of the formula (1) are important intermediates for the production of valuable vat dyes, pigments and disperse dyes.

The invention is illustrated by the following Exam75 ples in which the parts are by weight.

Example 1

16 parts of 1-nitro-2-methylanthraquinone are suspended in 140 parts of N-methyl-2-pyrrolidone. The suspension is heated to 85°C and, at this temp-80 erature, a solutionof 3.2 parts of hydrazine hydrate in 40 parts of N-methyl-2-pyrrolidone are added dropwise in the course of 20 minutes. The reaction mixture is then stirred for 5 minutes, cooled to room temperature and filtered. The filter residue is non-reacted starting material (3 parts). Water is added to the filtrate, which is then filtered. The residue is washed neutral with water. Yield: 12 parts of 3-

methyl-pyrazolanthrone of the formula

90

95 Example 2

10 parts of 1-nitro-2-ethylanthraquinone are suspended at room temperature in 50 parts of sulfolane. The suspension is heated to 55°C and, at this temperature, 4 parts of hydrazine, dissolved in 20 parts of sulfolane, are added dropwise in the course of 2 hours. The reaction mixture is stirred for 10 minutes and filtered after the addition of 200 parts of a mixture of ice and water. The filter cake is washed neutral with water. Yield: 9 parts of

105 3-ethylpyrazolanthrone of the formula

N-NH C₂H₅

110 Example 3

10 parts of 1-nitro-2-ethylanthraquinone are suspended at room temperature in 50 parts of N-methyl-2-pyrrolidone. The suspension is heated to 115 60°C and, at this temperature, 4 parts of hydrazine dydrate, dissolved in 20 parts of N-methyl-2-pyrrolidone, are added dropwise in the course of 2 hours. The reaction mixture is stirred for 10 minutes and filtered after the addition of 200 parts of mixture of ice and water. The filter cake is washed neutral with water. Yield: 8.5 parts of 3-ethylpyrazolanthrone of the formula (4).

10 parts of 1-nitro-2-ethylanthraquinone are suspended at room temperature in 60 parts of sulfolane.

Then 4 parts of hydrazine hydrate, dissolved in 20 parts of sulfolane. Then 4 parts of hydrazine hydrate, dissolved in 20 parts of sulfolane, are added dropwise to this suspension in the course of 2 hours. The

Example 5

10 parts of 1-nitro-2-ethylanthraquinone are suspended at room temperature in 60 parts of N-methyl-2-pyrrolidone. Then 4 parts of hydrazine 10 hydrate, dissolved in 20 parts of Nmethyl-2-pyrrolidone, are added dropwise to this suspension in the course of 2 hours. The reaction mixture is then stirred for 10 minutes and filtered after the addition of 200 parts of a mixture of ice and 15 water. The filter cake is washed neutral with water. Yield: 8.2 parts of 3-ethylpyrazolanthrone of the formula (4). Example 6

10 parts of 1-nitro-2-ethylanthraquinone are sus-20 pended at room temperature in 50 parts of dimethyl sulfoxide. The suspension is heated to 55°-60°C and, at this temperature, 4 parts of hydrazine hydrate, dissolved in 200 parts of dimethyl sulfoxide, are added dropwise in the course of 2 hours. The reac-25 tion mixture is then stirred for 10 minutes and filtered after the addition of 200 parts of a mixture of ice and water. The filter cake is washed neutral with water. Yield: 9 parts of 3-ethylpyrazolanthrone of the

formula (4). 30 Example 7

10 parts of 1-nitro-2-carboxyanthraquinone are suspended at room temperature in 50 parts ofsulfolane. The suspension is heated to 55°C and, at this temperature, 4 parts of hydrazine hydrate, dissolved 35 in 20 parts of sulfolane, are added dropwise in the course of 2 hours. The reaction mixture is stirred for 10 minutes and filtered after the addition of 200 parts of a mixture of ice and water. The filter cake is washed neutral with water. Yield: 9 parts of 40 3-carboxypyrazolanthrone of the formula

45

Example 8

10 parts of 1-nitro-2-carboxyanthraquinone are suspended at room temperature in 50 parts of N-methyl-2-pyrrolidone. The suspension is heated to 50 50 °C and, at this temperature, 4 parts of hydrazine hydrate, dissolved in 20 parts of Nmethyi-2-pyrrolidone, are added dropwise in the course of 2 hours. The reaction mixture is then stirred for 10 minutes and filtered after the addition of 55 200 parts of a mixture of ice and water. The filter cake is washed neutral with water. Yield: 8.5 parts of 3-carboxypyrazolanthrone of the formula (5). Example 9

10 parts of 1-nitro-2-carboxyanthraquinone are 60 suspended at room temperature in 60 parts of N-methyl-2-pyrrolidone. Then 4 parts of hydrazine hydrate, dissolved in 20 parts of Nmethyl-2-pyrrolidone, are added dropwise to this suspension in the course of 2 hours. The reaction

addition of 200 parts of a mixture of ice and water. The filter cake is washed neutral with water and dried. Yield 8.4 parts of 3-carboxypyrazolanthrone of the formula (5).

70 Example 10

10 parts of 1-nitro-2-carboxyanthraquinone are suspended at room temperature in 60 parts of sulfolane. Then 4 parts of hydrazine hydrate, dissolved in 20 parts of sulfolane, are added dropwise to this 75 suspension in the course of 2 hours. The reaction mixture is then stirred for 10 minutes and filtered after the addition of 200 parts of a mixture of ice and water. The filter cake is washed neutral with water. Yield: 8.6 partsof 3-carboxypyrazolanthrone of the 80 formula (5).

Example 11

10 parts of 1-nitro-2-carboxyanthraquinone are suspended at room temperature in 50 parts of dimethyl sulfoxide. The suspension is heated to 85 55°-60°C and at this temperature 4 parts of hydrazine hydrate, dissolved in 20 parts oof dimethyl sulfoxide, are added dropwise to this suspension in the ourse of 2 hours. The reaction mixture is then stirred for 10 minutes and filtered after the addition of 200 parts of 90 a mixture of ice and water. The filter cake is washe dneutral with water. Yield: 9 parts of 3-carboxypyrazolanthrone of the formula (5).

CLAIMS

95

1. A process for the production of 3-substituted pyrazolanthrones of the formula

wherein X is C₁-C₆alkyl, C₁-C₆alkoxy or a group -COR, in which R is hydrogen, C1-C4elkyl, hydroxyl, C1-105 C6alkoxy, -NH2, -NH-C1-C6alkyl or -NR1R2, in which R1 and R2, each independently of the other are C1-Csalkyl, which comprises reacting 1-nitroanthraquinones of the formula

wherein X is as defined in formula (1), in aprotic 115 dipolar solvents, with hydrazine or hydrazine hydride, at normal or slightly elevated temperature, to give 3-substituted pyrazolanthrones of the formula (1).

A process according to claim 2, wherein 120 1-nitro-2-methylanthraquinone, 1nitro-2-ethylanthraquinone or 1nitro-2-carboxyanthraquinone is reacted in N-methyl-2-pyrrolidone, sulfolane or dimethyl sulfoxide, with hydrazine hydrate, to give

125 3-methylpyrazolanthrone, 3-ethylpyrazolanthrone or 3-carboxypyrazolanthrone. 3. The 3-substituted pyrazolanthrones obtained

by the process according to claims 1 and 2.

4. A method of producing vat dyes, pigments

- 3-substituted pyrazolanthrones obtained by the process according to claims 1 and 2.
 - 5. A process according to claim 1 or 2 which is carried out at a temperature of 0 to 100 $^{\circ}$ C.
 - 6. A process according to claim 1 substuntially as hereinbefore described.

Printed for Her Majesty's Stationery Office, by Croydon Printing Company Limited, Croydon Surrey, 1978. Published by the Patent Office, 25 Southempton Buildings, London, WC2A 1AY, 'rom which copies may be obtained.